Quantum Material Synthesis by Reinforcement Learning

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Abstract

Designing conditions for experimental synthesis is the primary bottleneck for the realization of new functional quantum materials. Current strategies to synthesize new promising materials with desired properties are based upon a trial and error approach, which is a time consuming process and does not generalize to different materials. Here, we use deep reinforcement learning to learn synthesis schedules, which are time-dependent synthesis conditions of temperatures and reactant concentrations for a prototypical quantum material, monolayer MoS$_2$ via chemical vapor deposition (CVD). The reinforcement learning (RL) agent is coupled to a deep generative model that captures the probability density function of MoS$_2$-CVD dynamics and is trained on 10,000 computational synthesis simulations. After training, the RL agent successfully learns the optimal policy in terms of threshold temperatures and chemical potentials for the onset of chemical reactions and provides mechanistic insight to predict new synthesis schedules that produce well-sulfidized crystalline and phase-pure MoS$_2$ in minimum time, which is validated by reactive molecular dynamics.

1 Introduction

Advancement of technology based on promising new materials requires significantly shortening the current materials development timeline of $\sim$ 20 years [1]. This long timeline occurs due to the combination time needed to discover new candidate materials with desired set of properties from a vast search space and then identify scalable synthesis route for these materials. In recent years,
significant development has been made to address the first challenge using data-driven material science, which combines machine learning (ML) methods with first-principal based simulations for accelerated discovery of new materials. The exponential growth in computational power combined with high-throughput simulation helped us to build rich material database, which is mined by ML models to discover new material [2, 3]. Successful example of this strategies has created new ultrahard materials, battery materials, polymers, organic solar cells, OLEDs, thermoelectrics etc [4, 5, 6].

Another important component in rapid materials development is the identification of experimental synthesis route of these promising materials and composition, which has not kept pace with the computational materials screening [7, 8]. Current strategies for material synthesis are based on trial and error approach and are largely based on empirical insight and materials intuition. There have been limited attempts at predicting the outcome of chemical reactions for the solution synthesis of small molecules using chemical insights and machine learning [9, 10]. However, synthesis planning for bulk inorganic materials and non-solution based quantum material is still in its infancy as they involve complicated time-correlation between synthesis parameters [11, 12]. This requires considerably more refined models than previous efforts which only considered the combination of reactants to learn the outcome of chemical reaction of molecular and organic systems [13, 14]. Efforts based on text-mining of published literature on synthesis schedule has been made to understand the effect of solvent concentrations, heating temperatures, processing times, and precursors on synthesis schedules of materials [15, 16]. However, even these upcoming ML techniques are limited both by the scarcity and sparsity of data in terms of existing schedules and synthesized materials and therefore their extension to new, potentially unknown materials [15].

In this work, we describe a reinforcement learning (RL) [17] scheme to optimize synthesis routes for a prototypical member of the family of 2D quantum material, MoS\(_2\), via Chemical Vapor Deposition (CVD). CVD, a popular scalable technique for the synthesis of 2D materials, has numerous time-dependent parameters such as temperature, flow rates, concentration of gaseous reactants, and type of reaction precursors, dopants and substrates (together referred to as the synthesis profile) that need to be optimized [18]. Specifically, we use RL to (1) identify synthesis profiles that create material structure with desired properties, which in our case is maximum phase fraction of semiconducting crystalline phase of MoS\(_2\) in shortest possible time, (2) effect of different RL policies on the quality of generated MoS\(_2\) structure in terms of time dependent synthesis parameters and mechanistic insight of the synthesis process. Experimental synthesis is time-consuming and not suitable for high-throughput screening. Therefore, we have used reactive molecular dynamics (RMD) [19] to simulate CVD process that has previously shown to reproduce the potential energy surface of reacting system as well as capture important mechanisms of the CVD synthesis reaction [20, 21].

2 Method

2.1 Neural Autoregressive Density Estimation of CVD Dynamics

We perform RMD simulations to simulate synthesis of MoS\(_2\) by CVD with a multi-step reaction of MoO\(_3\) crystal in a sulfidizing atmosphere containing \(\text{H}_2\text{S}, \text{S}_2\text{, and H}_2\text{ gases}\). Each MD simulation
Figure 2: (a) Schematic of the NADE-CVD architecture consisting of fully connected networks as encoder ($F_{\text{MLP}_{\text{encoder}}}$) and decoder ($F_{\text{MLP}_{\text{decoder}}}$) and an intermediate LSTM block ($F_{\text{LSTM}}$). (b) Shows accuracy of NADE-CVD on 1000 RMD simulation. (c) Shows schematics of the RL framework for the synthesis schedule for MoS\textsubscript{2} synthesis.

models a 20-ns long synthesis schedule, divided into 20 steps, each 1 ns long and characterized by 4 variables: the system temperature, and the number of S\textsubscript{2}, H\textsubscript{2}S and H\textsubscript{2} molecules in the reacting environment as $(T, n_{H_2}, n_{S_2}, n_{H_2S})$. The final output structure (MoS\textsubscript{2} + MoO\textsubscript{3}−\textsubscript{x}) generated at the end is a non-trivial function of its synthesis schedule as shown in Figure 1. Here, each RMD simulation takes $\sim$ 3 days, and thus it won’t be possible to directly use them to train the RL agent. For this purpose, we have built a neural autoregressive density estimator (NADE-CVD) to approximate the probability density function of MoS\textsubscript{2}-CVD dynamics in RMD simulation [22, 23, 24]. The probability density function of CVD dynamics, $P(X,Z)$, is characterized by two sets of random variables, which are (1) the observed variables, $X = X_{1:t_{\text{max}}}$, given by the user defined synthesis condition and (2) the unobserved variable $Z = Z_{1:t_{\text{max}}}$ given by the time dependent phase fraction of 2H, 1T and defect phases in the MoO\textsubscript{x}S\textsubscript{y} surface. Using conditional independence between variables and chain rule, $P(X,Z)$ is written as following autoregressive function:

$$ P(Z|X,Z_1) = P(Z_2|Z_1,X_1) \ldots P(Z_{t+1}|Z_{1:t},X_{1:t}) \ldots P(Z_T|Z_{1:t_{\text{max}}-1},X_{1:t_{\text{max}}-1}) $$

where $X_t = (T_t, n_{H_2}^t, n_{S_2}^t, n_{H_2S}^t)$ and $Z_t = (n_{2H}^t, n_{1T}^t, n_{\text{defect}}^t)$. Here, NADE-CVD models each of the conditional probability, $P(Z_{t+1}|Z_{1:t},X_{1:t})$, as a Gaussian distribution and output its parameters: mean $\mu_{t+1} = (\mu_{2H}^{t+1}, \mu_{1T}^{t+1}, \mu_{\text{defect}}^{t+1})$ and variance $\sigma_{t+1} = (\sigma_{2H}^{t+1}, \sigma_{1T}^{t+1}, \sigma_{\text{defect}}^{t+1})$. The architecture of NADE-CVD is given in Figure 2a, which consists of an encoder, LSTM cell and a decoder. The NADE-CVD is trained using 10,000 RMD simulation profile of 20 ns each using maximum likelihood estimate. Figure 2b shows the prediction error by NADE-CVD on test dataset of 1000 RMD simulations.

2.2 Reinforcement Learning for MoS\textsubscript{2} synthesis

Our RL workflow consists of an RL agent coupled with the NADE-CVD that serves as an environment of CVD synthesis, Figure 2c. The RL agent ($\pi_\theta$) is represented using a fully connected neural network, where each episode of RL is of length 20 and difference between consecutive timestep is 1 ns that is equivalent to a 20 ns synthesis schedule. Each episode of RL starts from an arbitrary synthesis condition, $(T^0, S_2^0, H_2S^0)$, and a MoO\textsubscript{3} crystal, where the goal of the RL agent is to maximize...
We further note that this synthesis schedule captures a non-trivial mechanistic detail about phase evolution during synthesis – the nucleation of the 1T phase precedes the nucleation of the 2H crystal structure, which was also previously observed in MoS$_2$ (Ref. 20). We also observe that the nucleation of the 2H phase occurs at lower temperature (3000 K) as compared to that of 1T phase (3000 K), which is in agreement with the predictions from RL agent. This strategy is consistent with atomistic simulation-based predictions of optimal synthesis conditions (Ref. 20).

The above objective function is equivalent to proposing actions that convert the initial MoO$_3$ crystal into 2H-MoS$_2$ structure in least time. Closer inspection of these policies provides mechanistic insights into the CVD process and the effect of variations of synthesis condition on the quality of final structure. Figure 3a shows one such policy proposed by the RL agent, which consists of an early high-temperature ($>$3000 K) phase spanning the first 7-10 ns followed by annealing to an intermediate temperature ($<$3000 K) for the reminder of the synthesis profile. This strategy is consistent with atomistic simulation-based synthesis of MoS$_2$, where high temperature ($>$3000 K) is necessary for both the reduction of MoO$_3$ surface and its sulfidation, and the subsequent lower temperature ($<$3000 K) is necessary for enabling crystallization in the 2H structure, while continuing to promote residual sulfidation. Figure 3b shows the validation of this RL-generated profile using NADE-CVD and RMD simulation, which shows both the NADE-CVD prediction and the actual RMD simulation results are in close agreement. We further note that this synthesis schedule captures a non-trivial mechanistic detail about phase evolution during synthesis – the nucleation of the 1T phase precedes the nucleation of the 2H crystal structure, which was also previously observed in MoS$_2$ synthesis simulations (Ref. 20).

Another important phenomenon identified by RL agent is the effect of initial gas concentration on the quality of the final synthesized material. In Figure 3c, we compute the phase fraction of 2H phase in the synthesized MoS$_2$ product over the last 10 ns of the simulation for 3200 synthesis conditions proposed by the RL agent under different initial gas concentrations. Here, higher mean of the probability distribution function provides an indication of the extent of sulfidation as well as the quality of the final synthesized material. In Figure 3c, we compute the phase fraction of 2H phase in the synthesized MoS$_2$ product over the last 10 ns of the simulation for 3200 synthesis conditions proposed by the RL agent under different initial gas concentrations. Here, higher mean of the probability distribution function provides an indication of the extent of sulfidation as well as the quality of the final synthesized material.
time required to generate 2H phases. Figure 3c shows that RL agent recommends low concentration of \( S_2 \) at early stages (0-3 ns) of the synthesis, when the temperature is high. This partially evacuated synthesis atmosphere with low \( S_2 \) concentration promotes the evolution of oxygen from and self-reduction of the \( MoO_3 \) surface that helps to generate in a significantly higher 2H phase fraction in the synthesized product.

4 Conclusion

We have developed a reinforcement learning scheme for the predictive synthesis of two-dimensional \( MoS_2 \) monolayers using chemical vapor deposition. The RL model successfully proposed several new reaction schedules, i.e. time-dependent reaction conditions, to synthesize \( MoS_2 \) with maximum crystallinity and phase fraction of the 2H semiconducting structure. More importantly, the RL model also provides mechanistic insight into the material synthesis process and an understanding of the role of synthesis conditions (temperature, chemical environment) on the quality of the synthesized crystal. This RL scheme provides the first viable high-throughput approach to screening material synthesis conditions to tackle the as-yet unsolved problem of predictive synthesis of novel nanomaterials.

Broader Impact

While the RL scheme described here is implemented on simulated chemical vapor deposition, the scheme is highly generalizable to any other synthesis technique that relies on time-dependent reaction conditions, including sputtering, pulsed laser deposition, flame synthesis, vapor transport etc. The model can also be used to predict synthesis schedules to yield more complex products, including heterostructures containing interfaces between multiple phases. By modifying the reward function used in the RL model to capture functional properties (and not just structure/ phases) of the synthesized product, the RL scheme can be directly used to identify promising synthesis schedules to fabricate materials with desired properties. This provides an complementary framework to existing materials discovery paradigm based on high-throughput ab initio calculations.

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References


